Sai Kung Sung Tsun Catholic School (Secondary Section) F.5 Chemistry TEACHING SYLLABUS

	Topic	Experiment/ Activity	
1.	Microscopic World II	1. Building models of molecules	
	1.1 Shapes of molecules	2. Testing liquids to find out if their molecules are	
	1.2 Bond polarity and intermolecular forces	polar	
		3. Investigating the temperature changes caused by	
		the evaporation of liquids with different strength	
_		of intermolecular attractions	
2.	Fossil Fuels and Carbon Compounds	1. Investigating important reactions of alkanes	
	2.1 Fossil fuels	2. Cracking of medicinal paraffin and testing the	
	2.2 Homologous series, structural formulae	gaseous product	
	and naming of carbon compounds	3. Investigating the chemical properties of an alkene	
	2.3 Alkanes and alkenes		
	2.4 Addition polymers	• •	
		ision	
2		examination	
3.	Chemistry of Carbon Compounds	1. Oxidizing ethanol to ethanoic acid and testing the	
	3.1 An introduction to the chemistry of	ethanoic acid and produced	
	carbon compounds	2. Studying the reaction between ethanol and	
	3.2 Isomerism	ethanoic acid	
	3.3 Typical reactions of selected functional	3. Preparing and purifying	
	groups 3.4 Synthesis of carbon compounds	2-chloro-2-methylpropane 4. Preparing a soap and testing its properties	
	3.5 Important organic substances	4. Freparing a soap and testing its properties	
	· •	rm Test	
4.	Chemical Reactions and Energy	1. Finding pairs of chemicals that may be used to	
	4.1 Energy changes in chemical reactions	make heat-packs and cold-packs	
	4.2 The enthalpy change and internal energy		
	change of a system during a reaction	neutralization	
	4.3 Measuring enthalpy changes	3. Determining the strength of the hydrogen bond	
	4.4 Hess's Law	between ethanol molecules.	
	4.5 Use Hess's Law to determine enthalpy	4. Determining the enthalpy changes of combustion	
	changes	of some alcohols	
		5. Determining the enthalpy change of formation of	
		the magnesium oxide	
		6. Determining the enthalpy change of thermal	
		decomposition of potassium hydrogencarbonate	
Revision			
	Second term examination		

<u>F.5 CHEMISTRY ASSESSMENT SYSTEM</u>:

Term Result (100%) = Term exam (50%) + Course work(50%) Course work (100%) = Uniform test /Term Tests/Quizzes(50%) + homework(40%) + learning attitude (10%) Annual Result (100%) = 1^{st} Term Result (50%) + 2^{nd} Term Result (50%)